

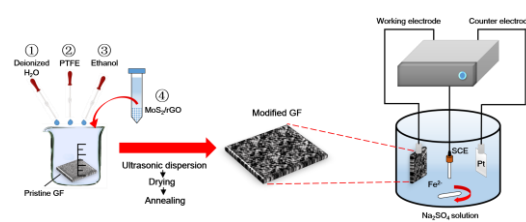
# Graphite felt incorporated with MoS<sub>2</sub>/rGO for electrochemical detoxification of high-arsenic fly ash

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**1. Introduction** – Accumulation of high-arsenic fly ash (HAFA) poses a serious environmental threat due to the toxicity. As(III) oxidation is a necessary step for HAFA detoxification and in situ H<sub>2</sub>O<sub>2</sub> electrogeneration via the two-electron-pathway (2e<sup>-</sup>) oxygen reduction reaction (ORR) is considered a promising technique for As(III) oxidation [1, 2]. MoS<sub>2</sub> has been reported to have attractive 2e<sup>-</sup> ORR catalytic activity [3]. However, there are still some problems to be solved: (i) its conductivity needs to be improved, (ii) gaseous O<sub>2</sub> is difficult to be effectively

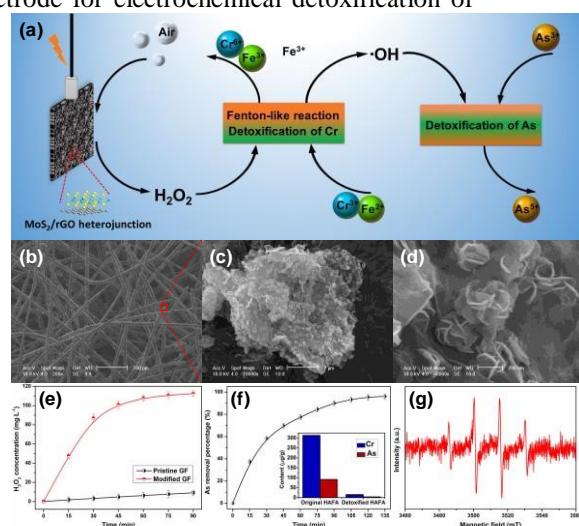


utilized, and (iii) the assembled electrode is not flexible and durable enough to treat solid waste.

**2. Experimental** - the experimental flow is schematically illustrated in Image 1.

**3. Results and Discussion** - A novel graphite felt (GF) modified with the MoS<sub>2</sub>/reduced graphene oxide (rGO) heterojunction is developed and demonstrated as an efficient electrode for electrochemical detoxification of

HAFA, as shown in Image 2(a). The modified GF has abundant microchannels (Image 2(b)), desirable surface properties, and fast electron transfer capability thereby enabling effective utilization of both dissolved O<sub>2</sub> and gaseous O<sub>2</sub> in 2e<sup>-</sup> ORR. The p-n junction consists of the p-type rGO substrate and uniformly distributed n-type MoS<sub>2</sub> nanoflowers, seeing Images 2(c) and 2(d). Theoretical calculation indicates that gaseous O<sub>2</sub> adsorb stably on sulfur vacancies and is reduced by electrons transferred from rGO. The modified GF exhibits superior ORR catalytic activity such as a high H<sub>2</sub>O<sub>2</sub> yield of 47.53 mg·L<sup>-1</sup>·h<sup>-1</sup>·cm<sup>-1</sup> (Image 2(e)) and onset potential of -0.12 V vs. SCE. The



·OH generated by the autocatalytic mechanism promotes oxidative dissolution of As(III) during detoxification of HAFA and the As removal percentage is 96.10% after electrolysis for 135 min, displayed in Images 2(f) and 2(g). The modified GF with excellent stability and durability has large industrial

potential in detoxification of HAFA and other As-bearing hazardous wastes.

**4. Conclusions** - This study makes significant contributions. Firstly, the gaseous O<sub>2</sub> reaction path in ORR is confirmed by the

combination of experiment and calculation. Secondly, 2e<sup>-</sup> ORR is innovatively applied to detoxification of high-arsenic solid waste.

## 5. References

- [1] M. Sun et al., *Environ. Sci. Technol.* 49 (2015) p. 9289–9297.
- [2] H.A. Maitlo et al., *Ind. Eng. Chem.* 73 (2019) p. 205–213.
- [3] Y. Xue et al., *Electrochim. Acta* 252 (2017) p. 245–253.

**Image 2.** (a) Graphical abstract of this study, (b) SEM image of the modified GF, (c) Low-magnification and (d) High-magnification SEM images of MoS<sub>2</sub>/rGO heterojunction, (e) H<sub>2</sub>O<sub>2</sub> accumulation versus time, (f) As removal percentage of HAFA versus time for the modified electrode (inset showing Cr and As contents in the original HAFA and detoxified HAFA), and (g) ESR spectrum of the hydroxyl radicals detected during the HAFA treatment.